

C Solek^b, A Rodríguez-Prieto^a^a Department of Organic Chemistry, UAM, Francisco Tomás y Valiente 7, Madrid, Spain^b Department of Manufacturing Engineering, UNED (Madrid – Spain)

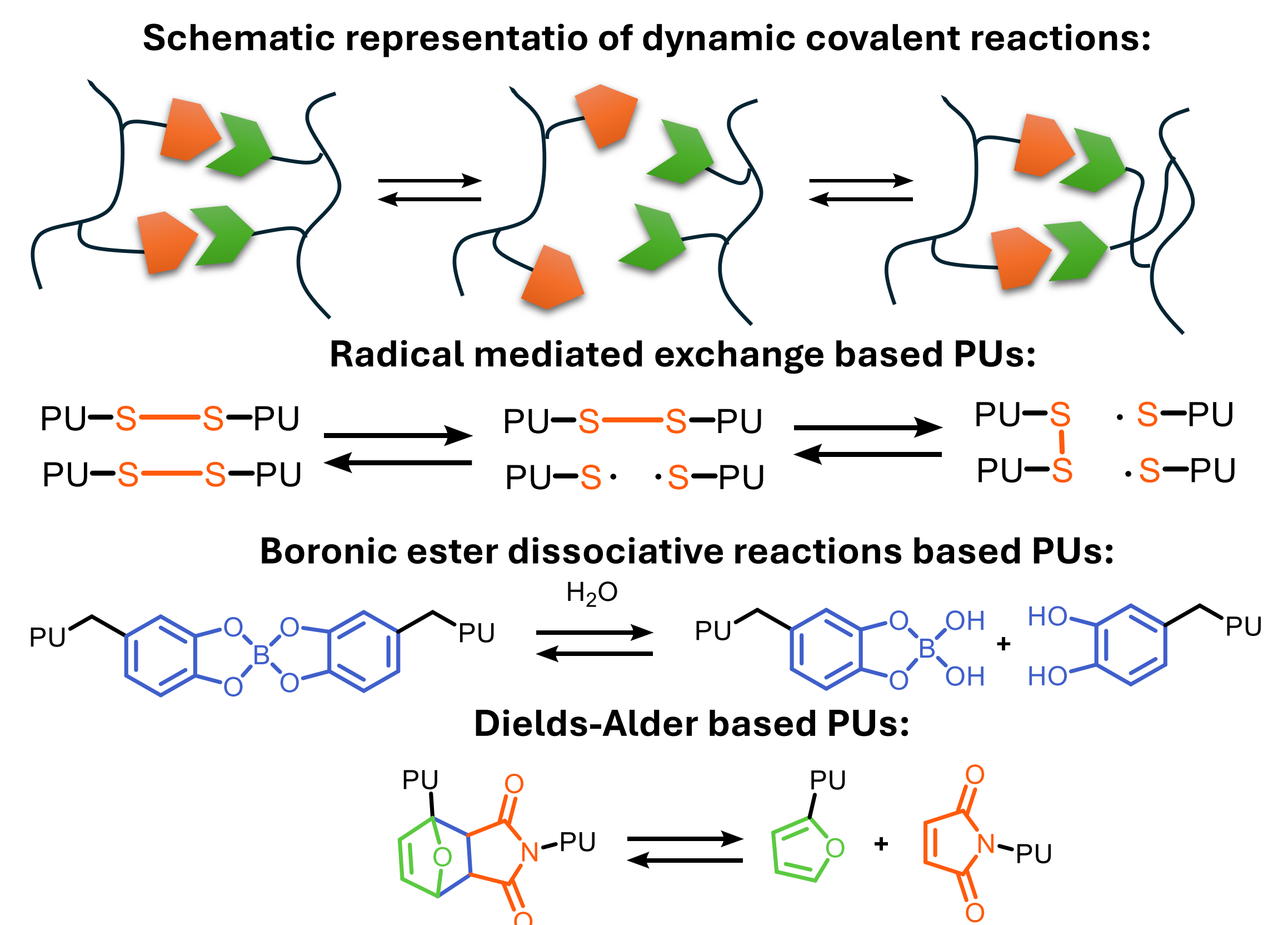
Our work is framed on the analysis of recent advances in self-healing polyurethanes, focusing on the incorporation of dynamic covalent and supramolecular bonds that allow self-repair under mild stimuli, significantly improving their mechanical properties and repair efficiency¹⁻⁴. The combination of covalent and non-covalent bonds offers a promising synergy, enhancing the balance between mechanical properties and self-healing capability^{2,3}. However, challenges in synthesis and production costs remain, driving the search for more sustainable methods.

Covalent Approach

Disulfide / Diselenide Bonds: can reform after breaking, allowing self-healing at room temperature under UV irradiation, achieving 100% recovery in tensile strength. Nonetheless, are sensitive to oxidizing agents, which can lead to degradation in harsh conditions.

Boroxine Bonds: form strong yet reversible links, enhancing self-healing capabilities, but require complex synthesis and specific conditions to ensure stability.

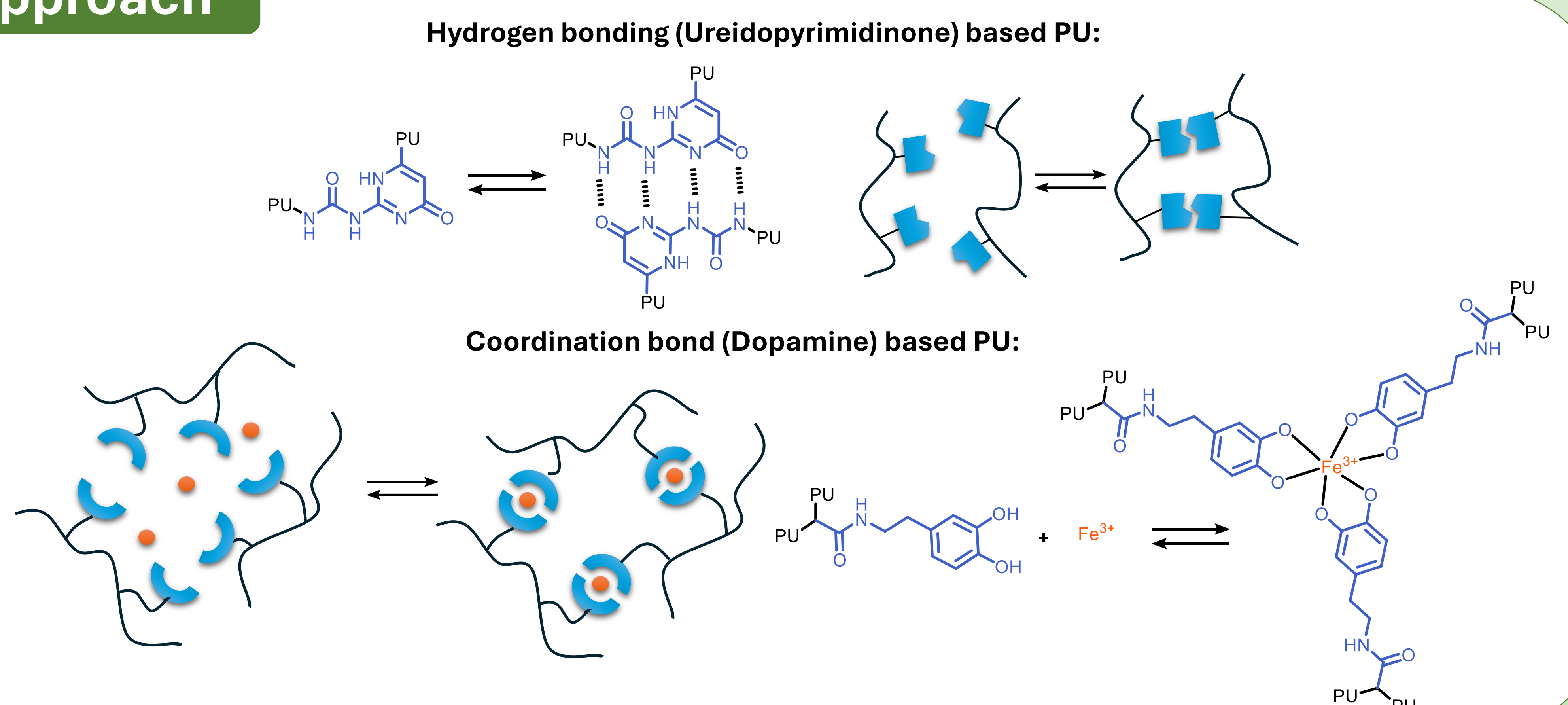
Diels-Alder Reactions: enable provide reversible covalent bonds for robust mechanical properties but are limited in uncontrolled environments.



Supramolecular Approach

Hydrogen Bonds: provide a quick and reversible self-healing, allowing materials to recovery rapidly from damage, but have lower mechanical strength, limiting material's overall durability.

Metal Coordination: provide high structural integrity and can be tuned for specific applications. However, the cost and availability of suitable metal ions can be a limiting factor.



Combination Approach

Disulfide & Hydrogen Bonds: demonstrate improved repair efficiency due to the rapid recombination of hydrogen bonds and the robust mechanical properties provided by disulfide bonds. Optimal performance requires precise synthesis control to ensure both bond function effectively.

Imine & Metal Bonds: enhances the mechanical strength and self-healing capabilities of the material, allowing it to recover from significant mechanical damage while maintaining structural integrity. However, synthesizing such system can be complex and expensive.

Challenges

Practical Implementation: the main challenge is the complexity of synthesis and the cost of functional monomers. Developing simpler and more economical methods for producing self-healing polyurethanes is crucial, as current synthesis processes often require complex steps and costly monomers, limiting commercial viability.

Conclusions

- Self-healing polyurethanes represent a significant advance in material science, offering substantial benefits in durability and sustainability.
- The integration of dynamic covalent and supramolecular bonds has shown promise in enhancing the equilibrium between mechanical properties and repair efficiency of these polymers.
- However, achieving practical implementation at an industrial scale will require overcoming several challenges.

ACKNOWLEDGEMENTS:

This work has been developed in the frame of the Projects PLEC2021-007750 (financed by MCIN/AEI/10.13039/501100011033 and by the European Union NextGenerationEU/ PRTR) and PID2022-143329OA-I00 (financed by MCIN/AEI).

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